

The influence of temperature on the spectral emittance of ash deposits taken from a 1.5 MW, pulverized coal test facility

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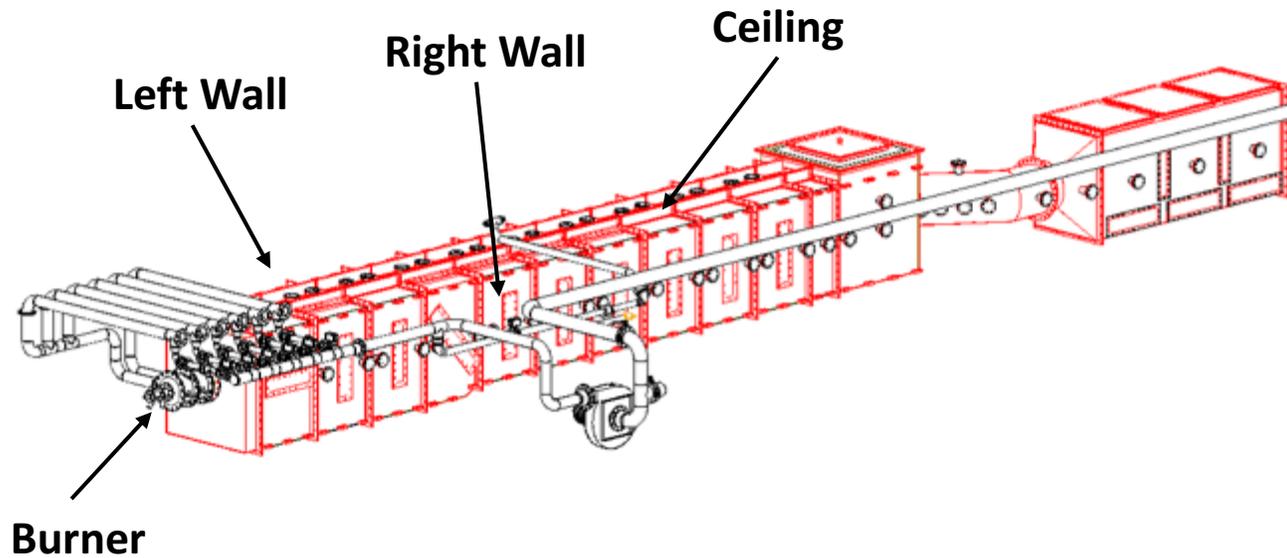
Introduction

- This work is part of the DOE-sponsored Carbon-Capture Multidisciplinary Simulation Center (CCSMC).
 - Overall CCSMC goal:
 - Create a predictive model of an industrial-scale, high efficiency, advanced ultra-supercritical oxy-coal fired power boiler.
 - One difficulty:
 - Deposits on the interior of the coal boilers significantly affect the heat transfer from the flame to the working fluid.
 - Deposit emittance can vary significantly over the following parameters:
 - Surface temperature
 - Microscopic structure/chemical composition
 - Macroscopic structure/surface morphology
 - Objective of this work:
 - Measure high-temperature emittance data from deposits in a 1.5 MW, pulverized-coal, oxy-combustion furnace (L1500 furnace)



Ash deposits in the L1500 furnace.

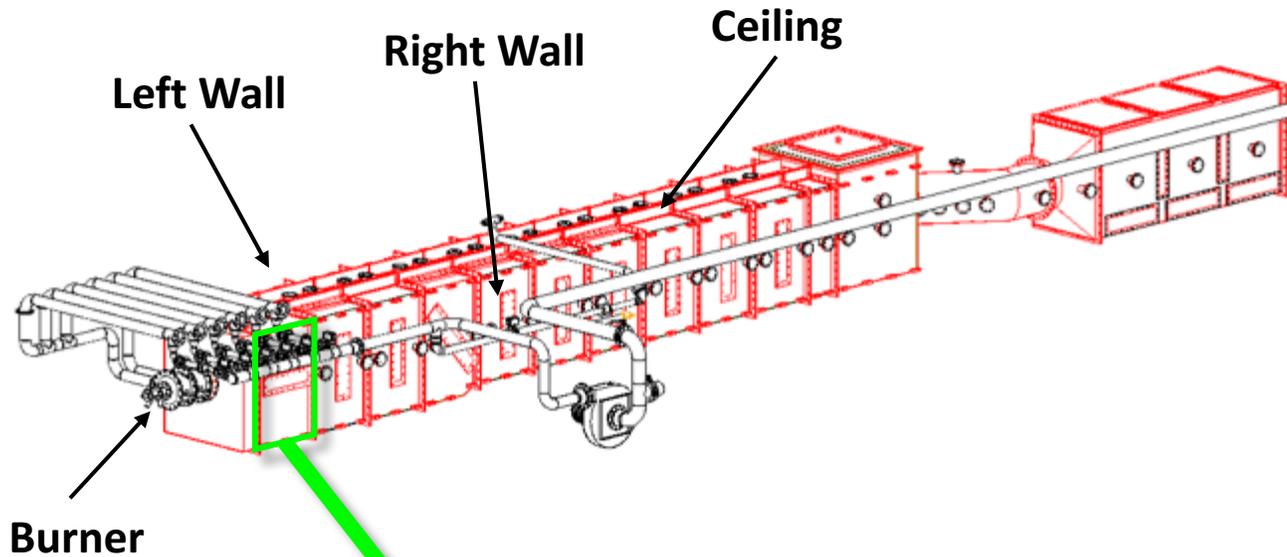
Sample Collection



L1500 furnace (1.1 m x 1.1 m cross section, 13.1 m in length)

- 396 samples were collected from the L1500 interior in a 1 ft x 1 ft grid
 - Surfaces: left wall, ceiling, & right wall

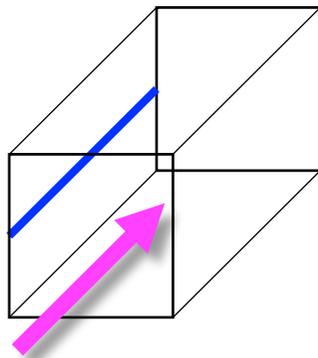
Sample Collection



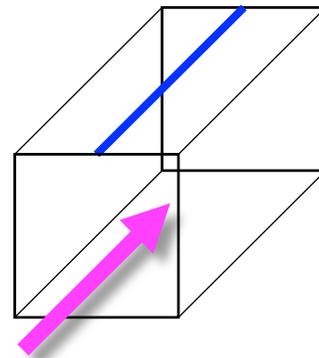
- 396 samples were collected from the L1500 interior in a 1 ft x 1 ft grid
 - Surfaces: left wall, ceiling, & right wall
- **Five** samples were chosen to be analyzed for emittance at high temperature (up to 1000 °C)
- All five samples were from the first section of the furnace (within 4 ft of the burner)

■ = sampling location
■ = flame

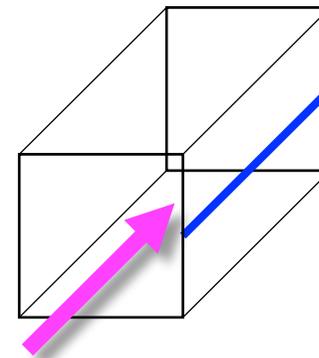
Left Wall



Ceiling



Right Wall

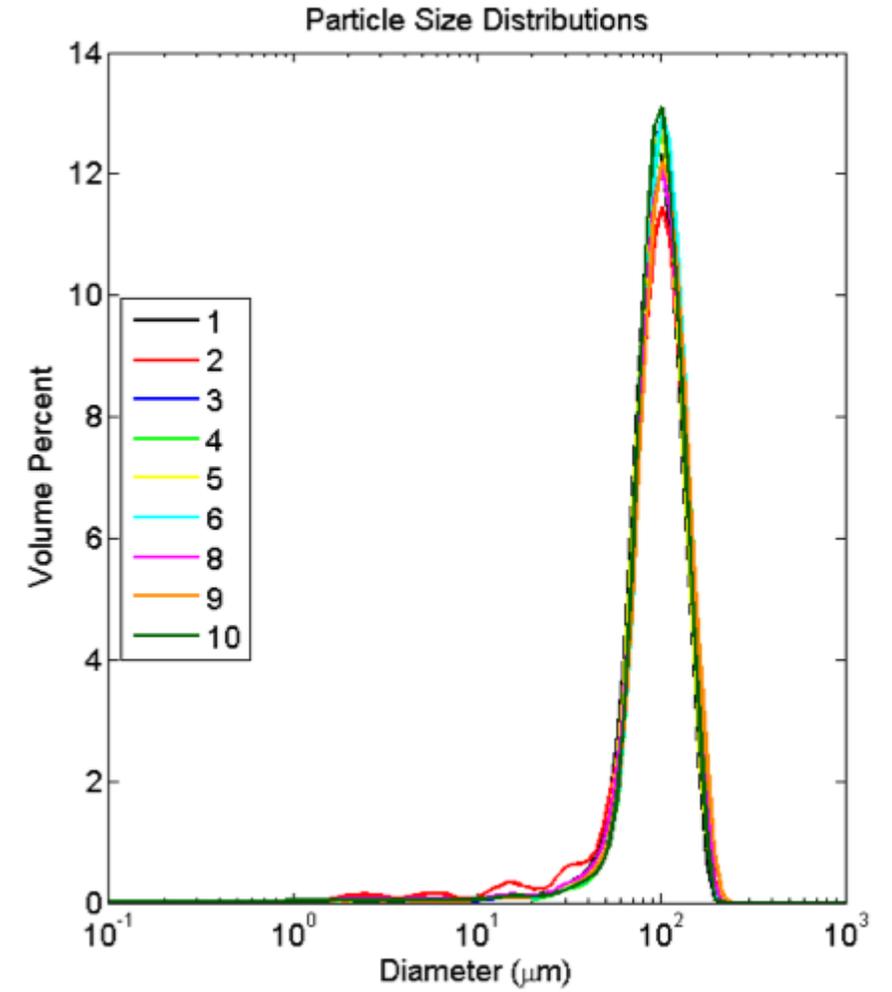
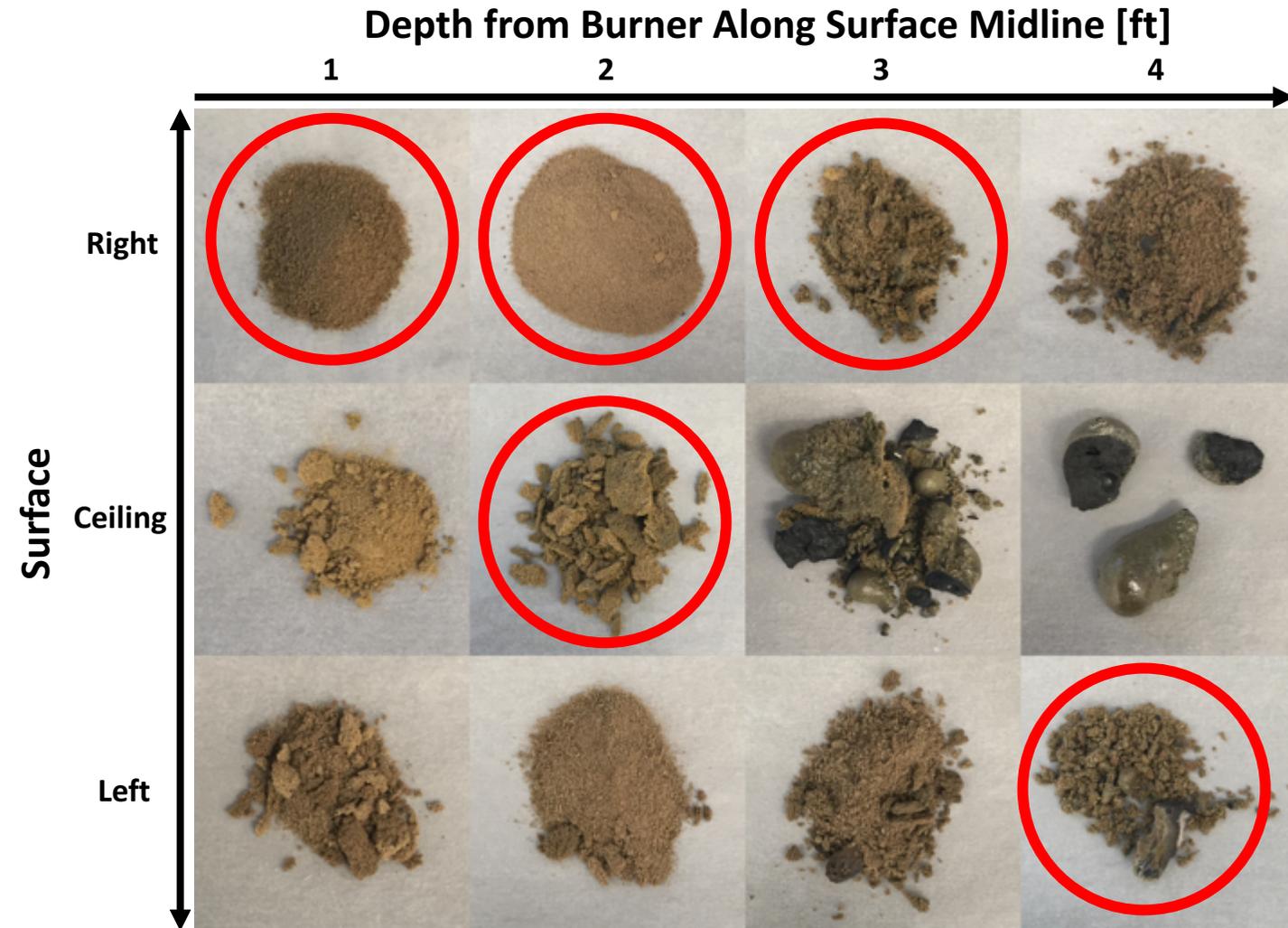


Sample Summary

- 10 measurements were taken
 - Nine measurements were ground and sieved to the same particle size distribution
 - One sample was a solid piece of a slag
 - Five sample locations examined (some of the measurements were to produce replicates)

Name	Sample #	Repetition	PSD (μm)	Surface	Depth (feet)
1v1	1	1	powder	Right	1
1v2	1	2	powder		
2v1	2	1	powder	Right	2
2v2	2	2	powder		
3v1	3	1	powder	Right	3
6v1	6	1	powder	Ceiling	2
6v2	6	2	powder		
6v3	6	3	powder		
10v1	10	1	powder	Left	4
10sv1	10s	1	solid	Left	4

Sample Preparation



- Samples were ground and sieved so that all would have the same particle size distribution.
 - NOTE: The sample images were taken before grinding and sieving.
- The red circles represent samples measured with high temperature emittance rig.

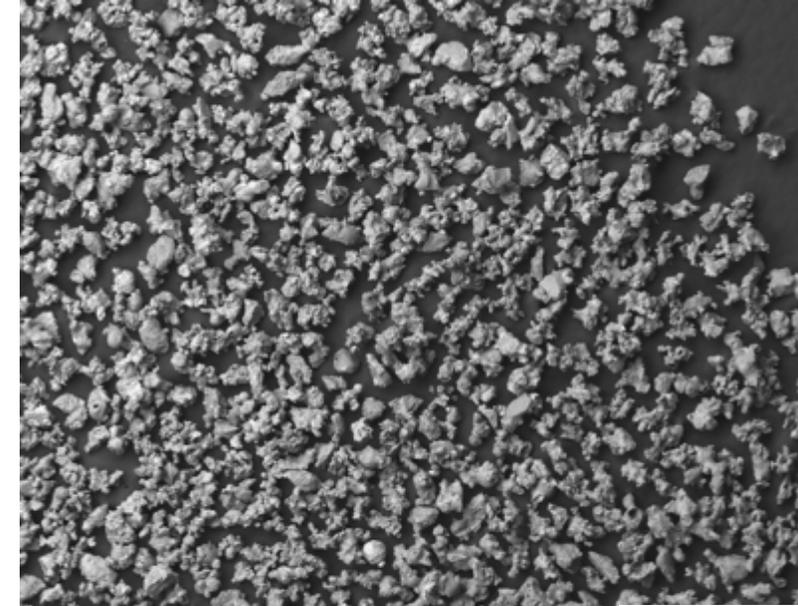
SEM

50x magnification



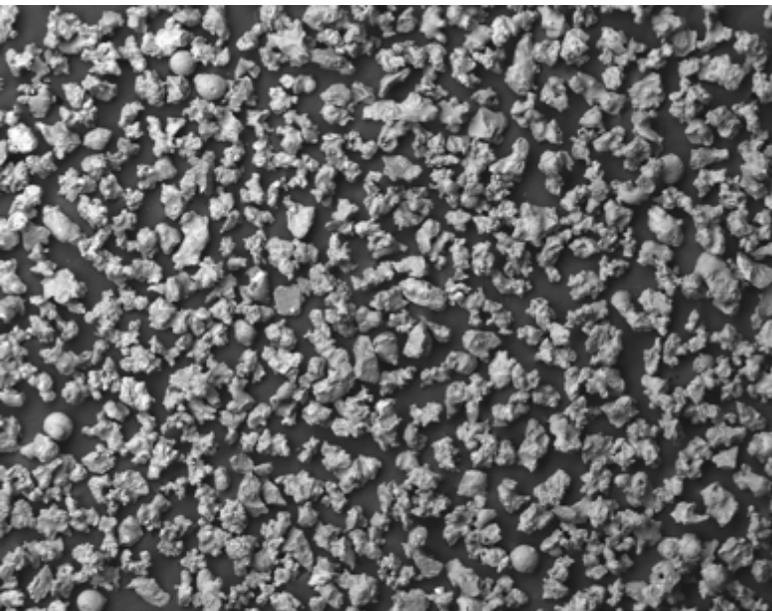
Sample 1

6/27/2016	mag	HV	det	WD	spot	500 μm
2:49:43 PM	50 x	10.00 kV	BSED	10.8 mm	5.0	1



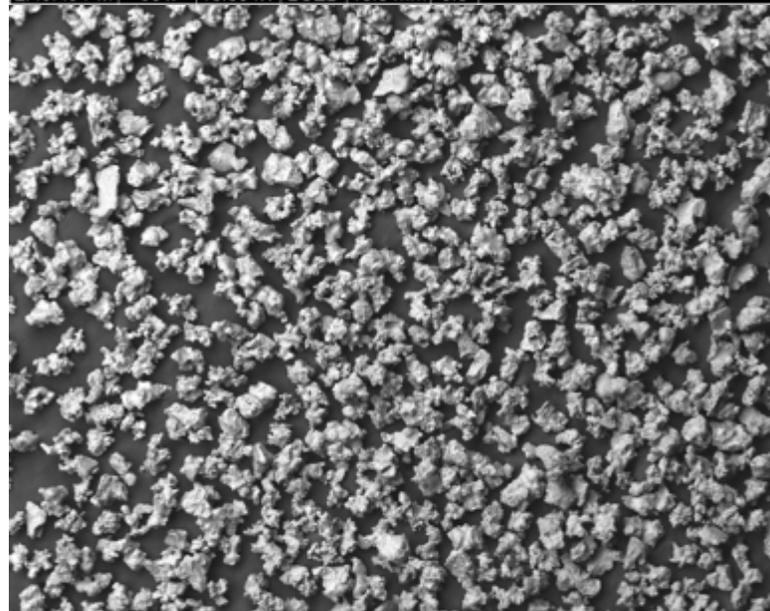
Sample 2

6/27/2016	mag	HV	det	WD	spot	500 μm
3:05:48 PM	50 x	10.00 kV	BSED	10.5 mm	5.0	2



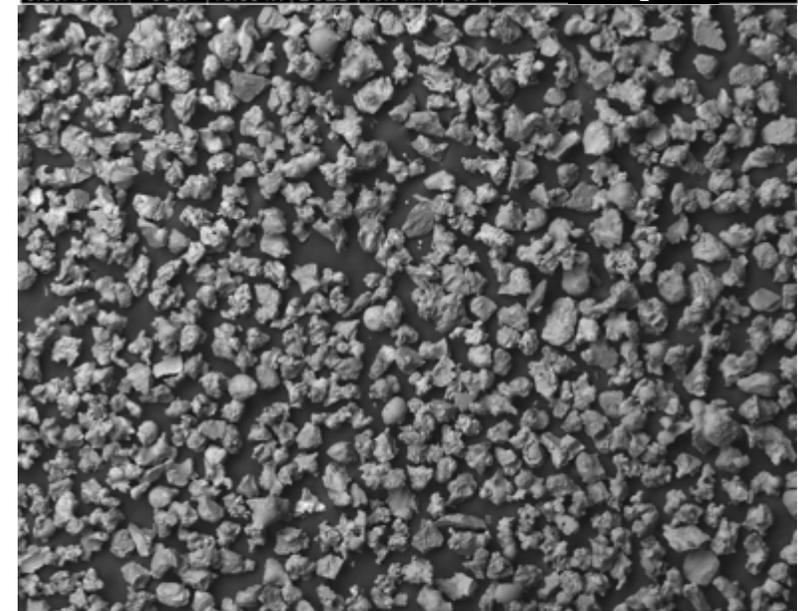
Sample 3

6/27/2016	mag	HV	det	WD	spot	500 μm
3:15:39 PM	50 x	10.00 kV	BSED	10.6 mm	5.0	3



Sample 6

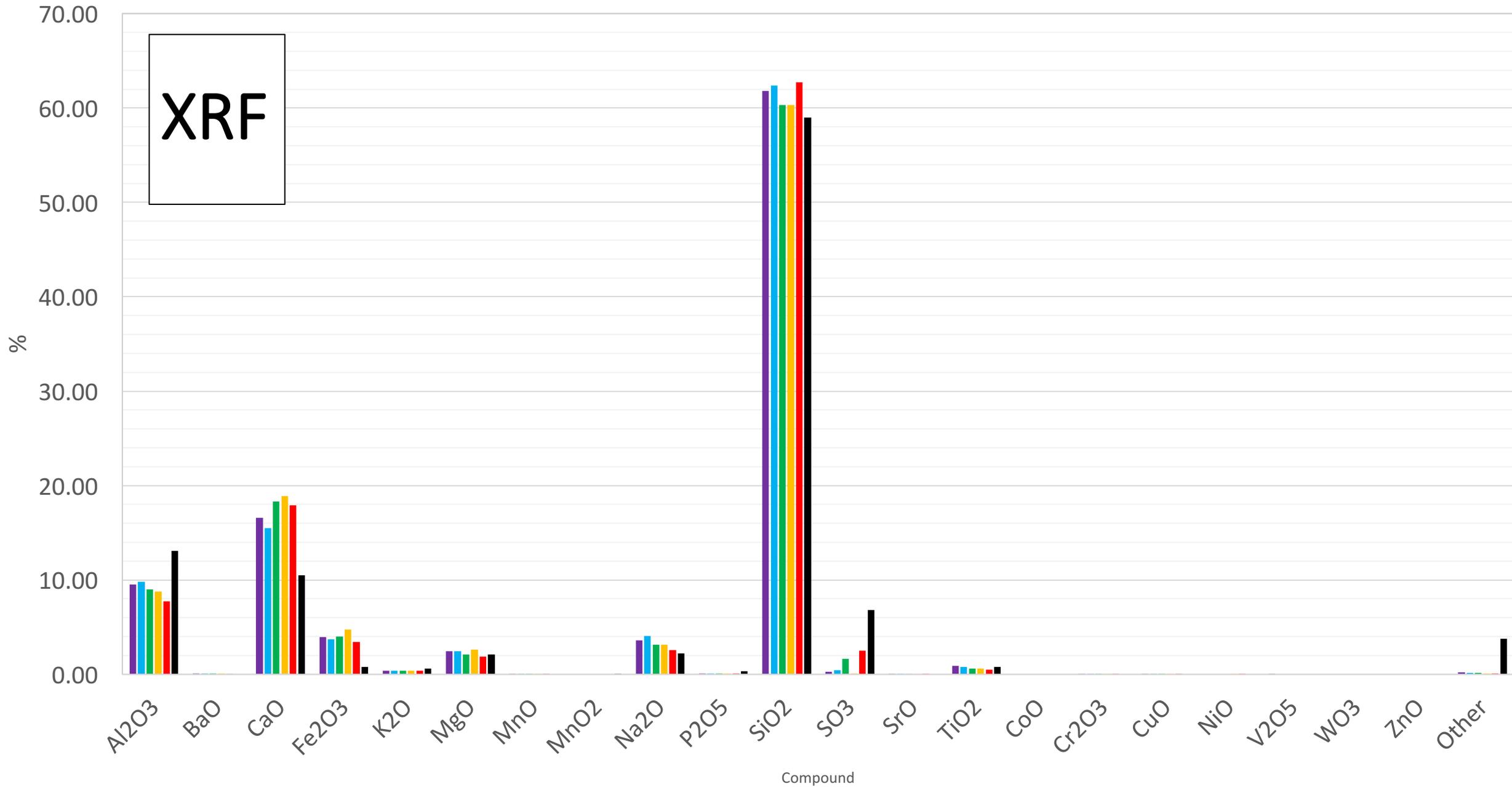
6/29/2016	mag	HV	det	WD	spot	500 μm
9:28:15 AM	50 x	10.00 kV	BSED	10.5 mm	5.0	6



Sample 10

6/29/2016	mag	HV	det	WD	spot	500 μm
9:42:45 AM	50 x	10.00 kV	BSED	10.6 mm	5.0	10

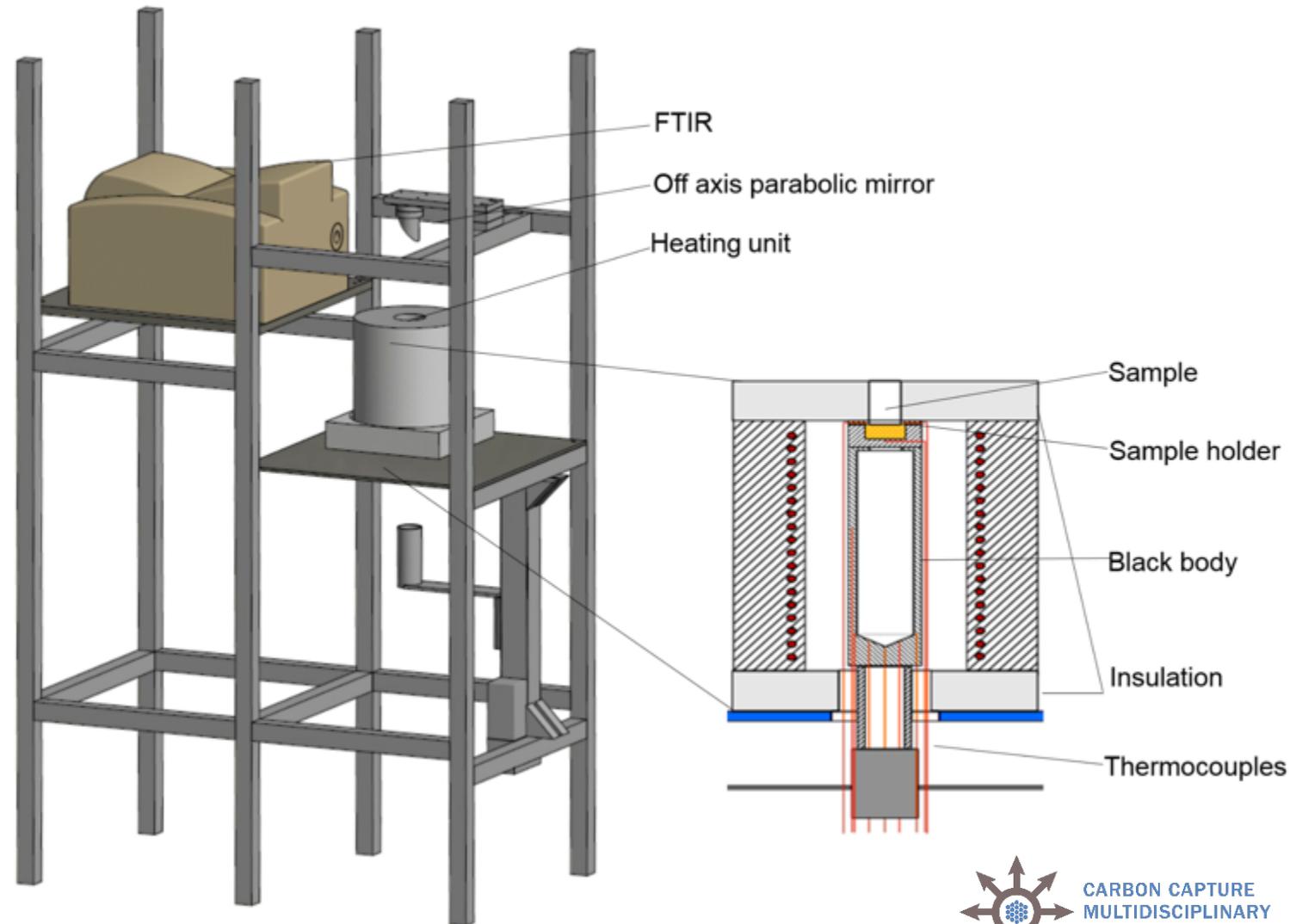
XRF



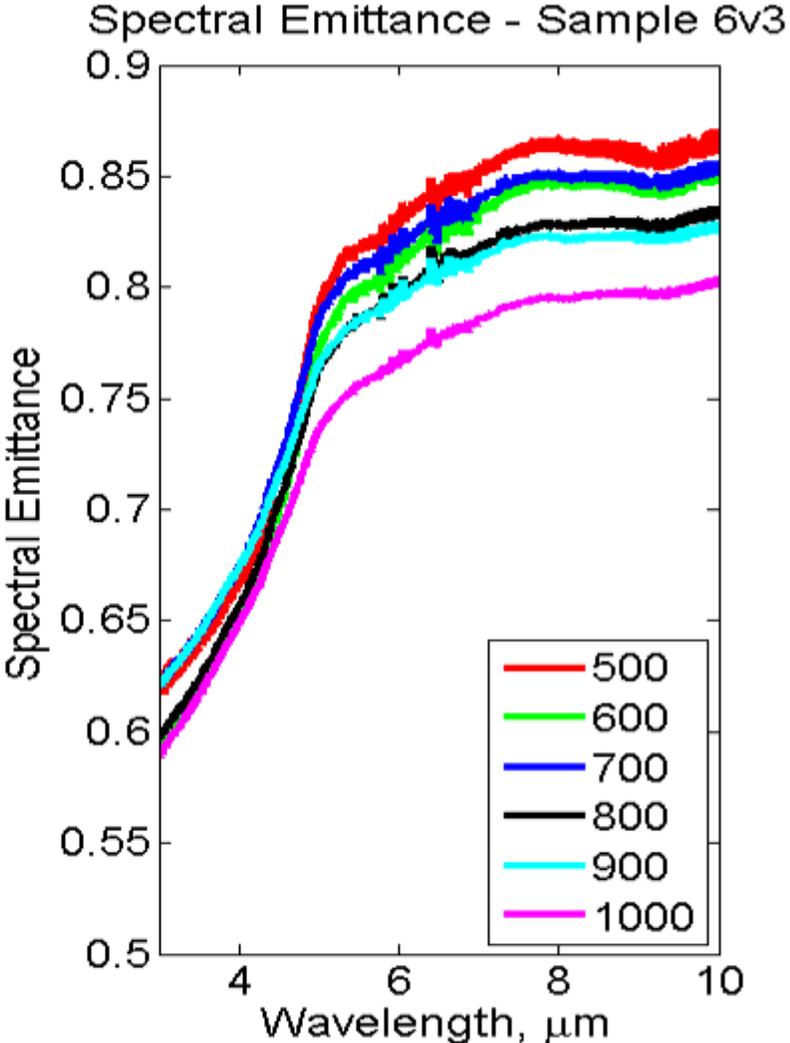
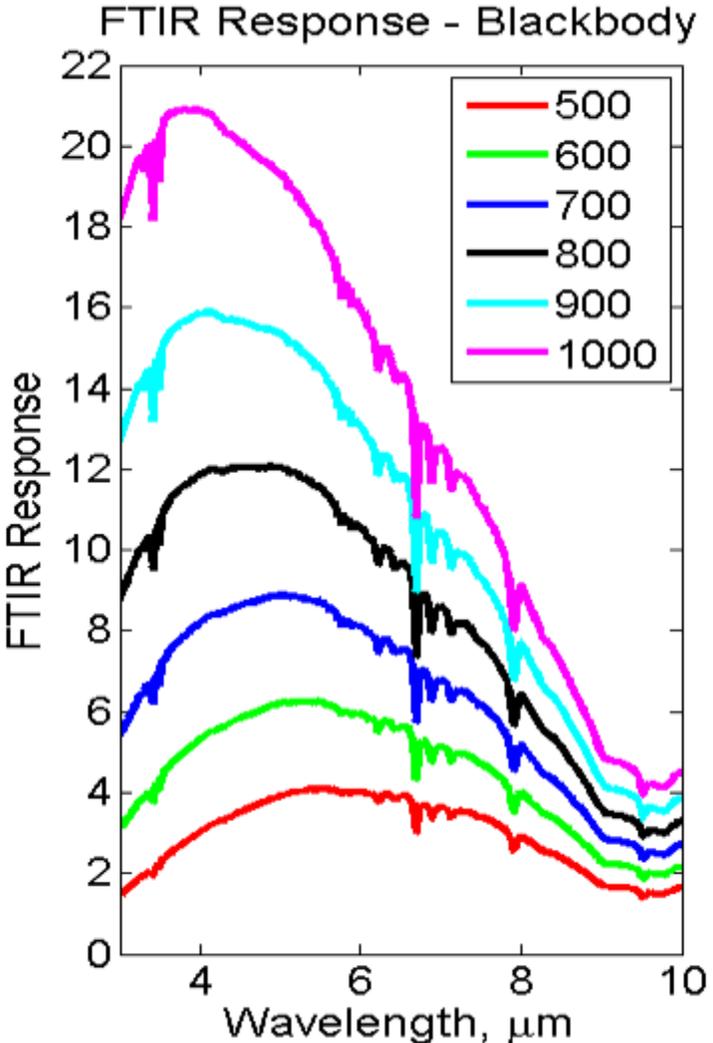
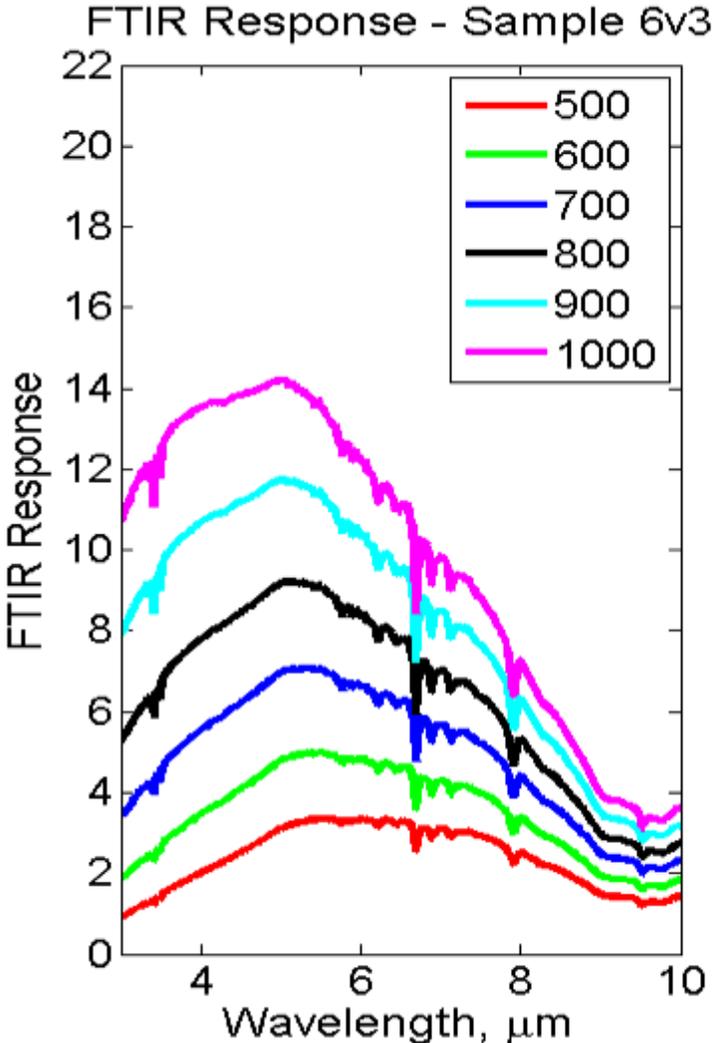
Sample 1 Sample 2 Sample 3 Sample 6 Sample 10 Sufco Mineral Analysis

Experimental Setup

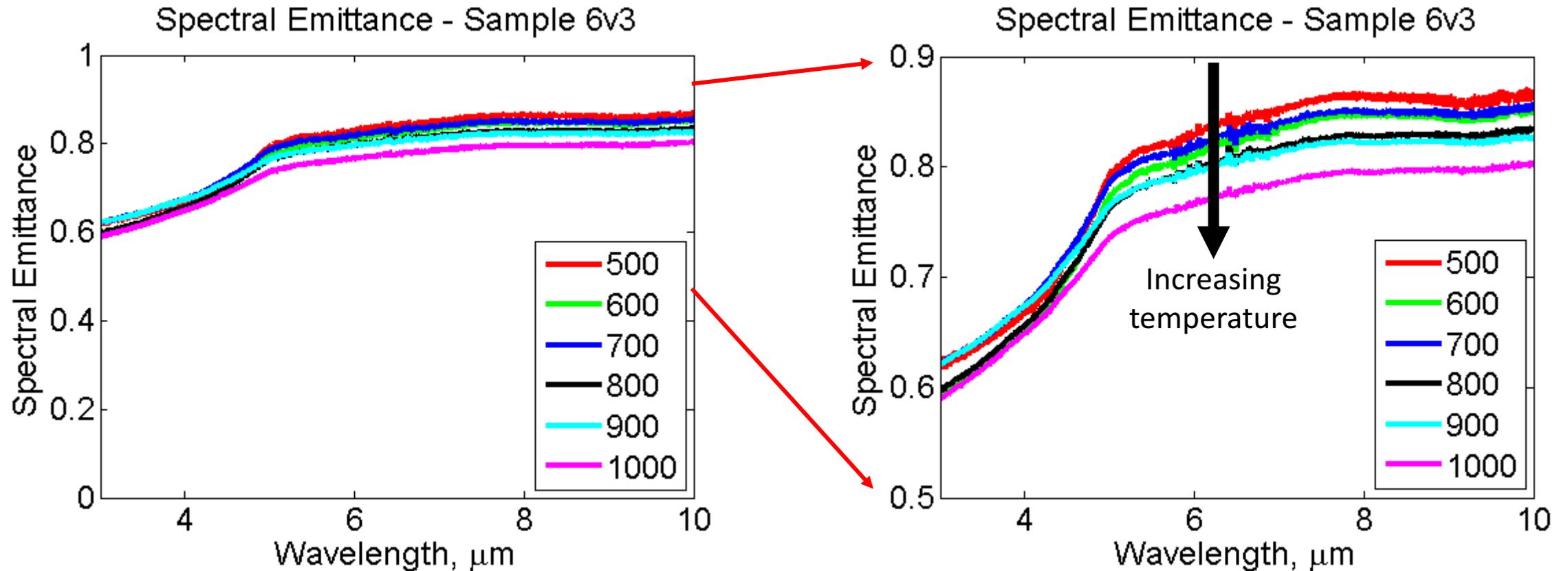
- The radiation test rig located at Ruhr University in Bochum, Germany was used to perform the high temperature emittance measurements (from 500-1000 °C and 0.68-28.5 μm).
- Radiation from the sample inside the rig, $L_S(\lambda, T)$, is directed into and measured with an FTIR.
 - $L_{S,\lambda}(T) \propto E_{\lambda}(T)$
- Radiation from a blackbody cavity inside the rig, $L_{BB}(\lambda, T)$, is also measured.
 - $L_{b,\lambda}(T) \propto E_{\lambda,b}(T)$
- The ratio of the two FTIR measurements is the spectral emittance.
 - $\varepsilon_{\lambda}(T) = \frac{E_{\lambda}(T)}{E_{\lambda,b}(T)} = \frac{L_{S,\lambda}(T)}{L_{b,\lambda}(T)}$
- Emittance vs. emissivity



Conversion from FTIR response to spectral emittance

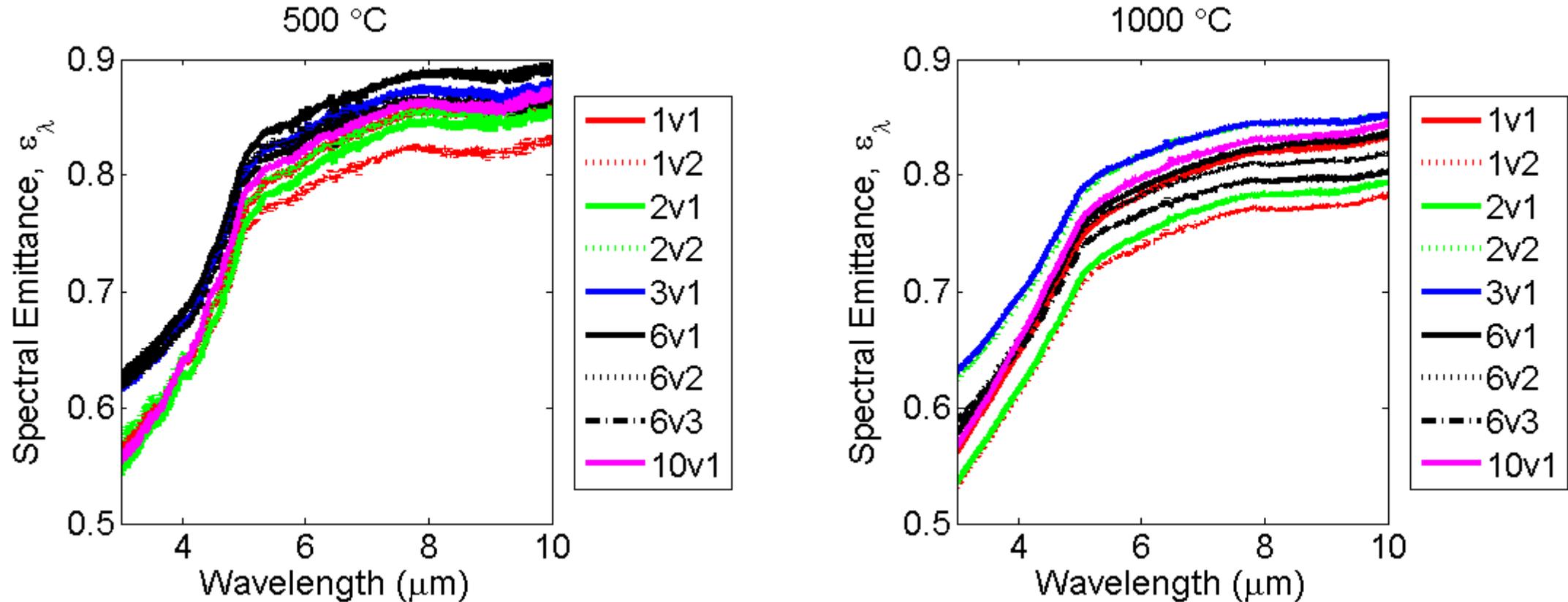


Spectral emittance as function of temperature



- Spectral emittance doesn't change significantly with temperature
- In general, the spectral emittance decreases with increasing temperature

Spectral Emittance for All Powdery Samples



- Spectral emittance for all powdery samples and their replicates
- Spectral emittance at each temperature for all powdery samples is fairly similar
 - Expected given the similarity in the compositions and particle size distributions
- The expected decrease in spectral emittance with increasing temperature is seen

Total emittance as function with temperature

- Total emittance calculation:

$$\bullet \quad \varepsilon_t(T) = \frac{\int_0^{\infty} \varepsilon_{\lambda}(T) E_{\lambda,b}(T) d\lambda}{\int_0^{\infty} E_{\lambda,b}(T) d\lambda}$$

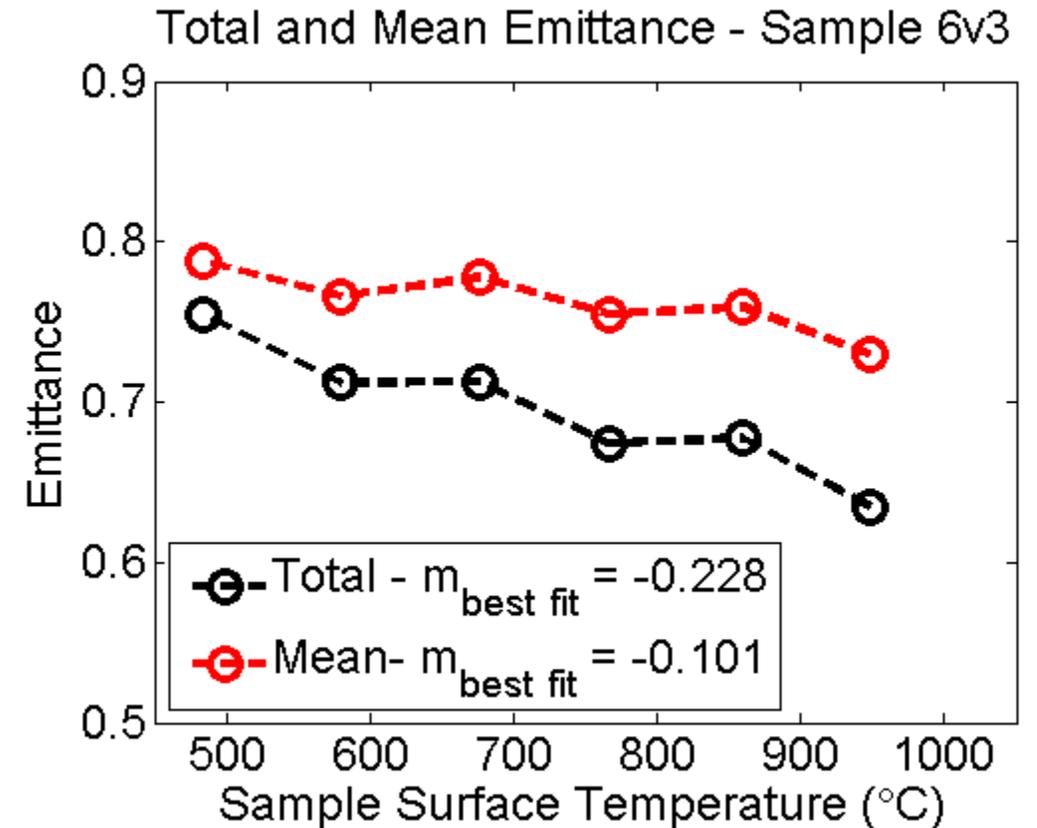
- Our signal was not strong enough below 3 μm or above 10 μm , so an approximation of the total emittance was calculated:

$$\bullet \quad \varepsilon'_t(T) = \frac{\int_{3\mu\text{m}}^{10\mu\text{m}} \varepsilon_{\lambda}(T) E_{\lambda,b}(T) d\lambda}{\int_{3\mu\text{m}}^{10\mu\text{m}} E_{\lambda,b}(T) d\lambda}$$

- In order to distinguish the contribution to the total emittance from changes in the spectral emittance versus changes in Planck's distribution (whose maximum changes as a function of temperature, a "mean emittance" is also plotted:

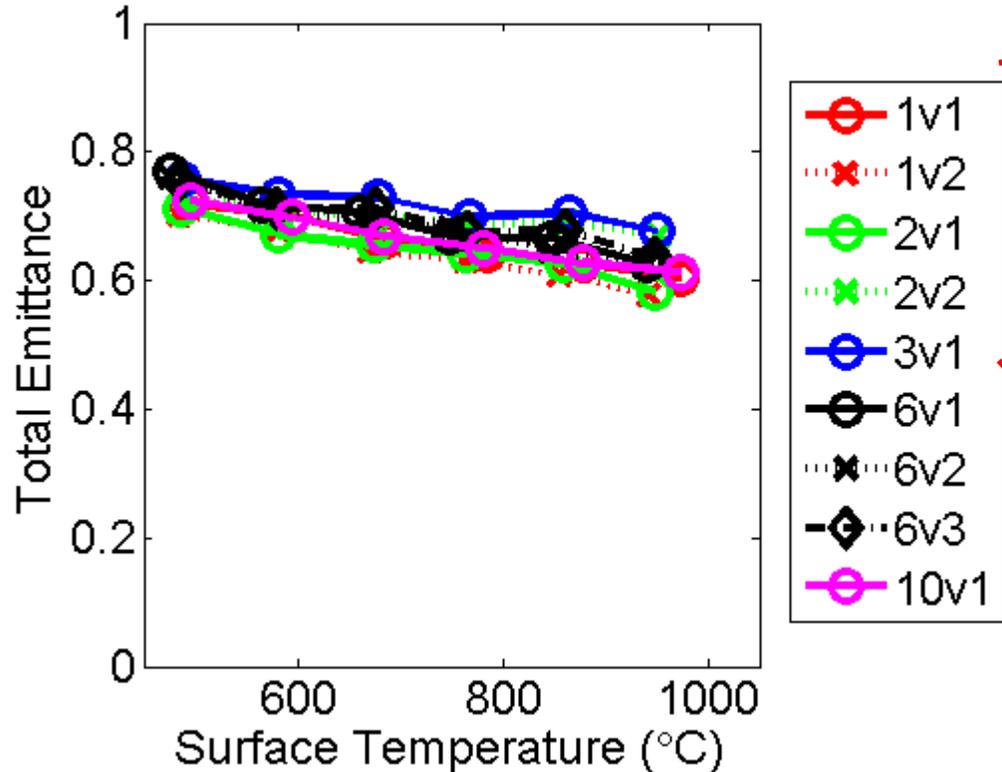
$$\bullet \quad \varepsilon'_m(T) = \text{average}(\varepsilon_{\lambda}(T))$$

- A downward trend in emittance with temperature is more dramatic for total emittance.
- Thus, take care when making conclusions about spectral emittance from total emittance

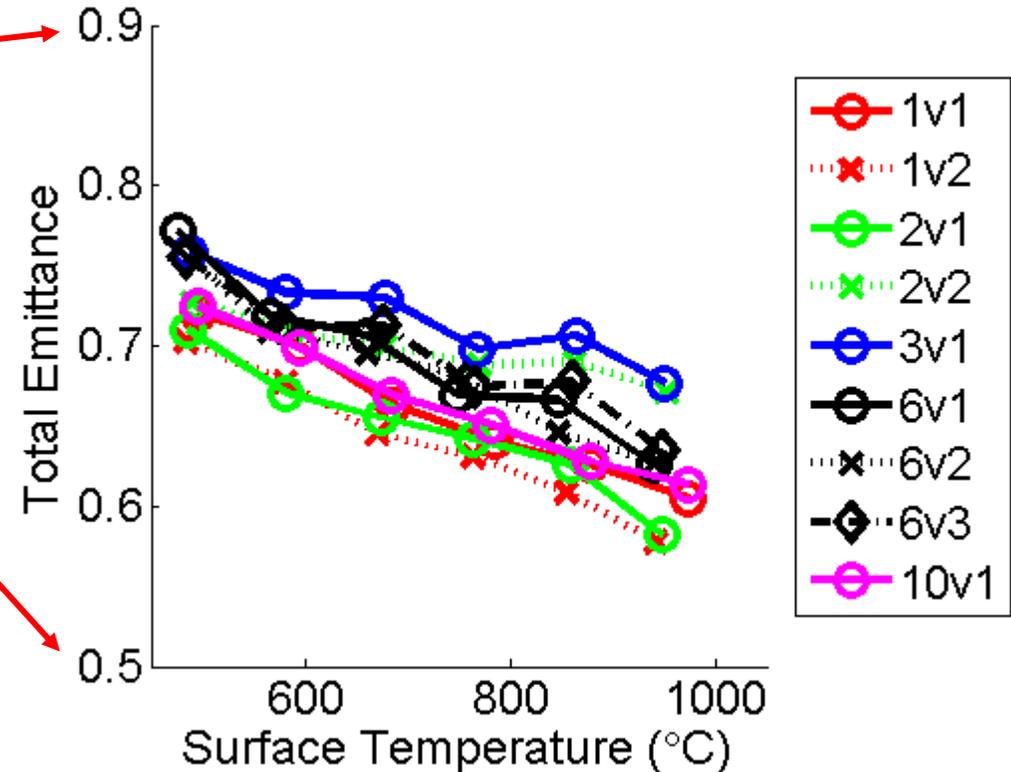


Total Emittance for All Powdery Samples

Total Emittance for Powdery Samples



Total Emittance for Powdery Samples



- Total emittance for all powdery samples is fairly similar
 - Expected given the similarity in the spectral emittances

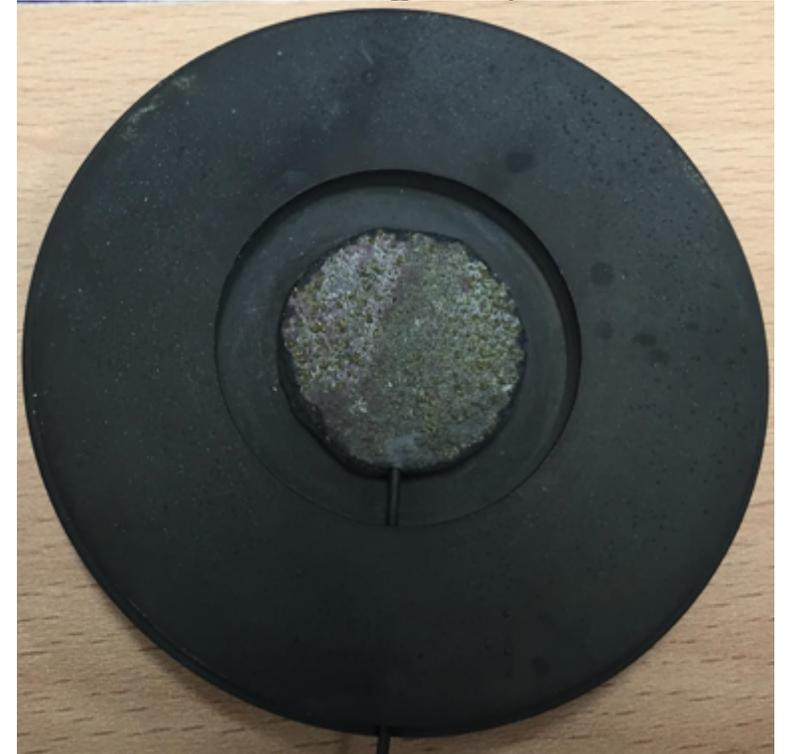
Effect of surface structure: Powder vs. Solid

- Only one sample contained a piece of slag large enough to be machined to fit the sample holder
- Smaller pieces of the slag from the sample were ground and sieved in the same procedure as the other powders
- This measurement isolates the effect of surface structure and temperature since the composition of two samples were identical

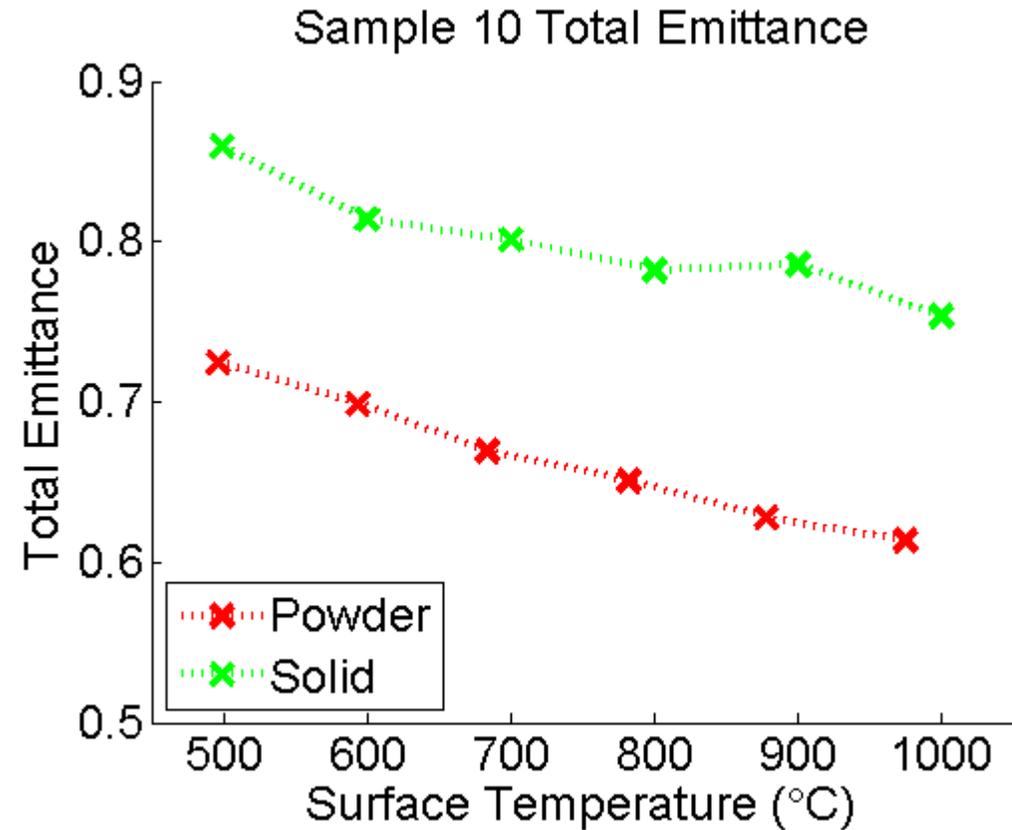
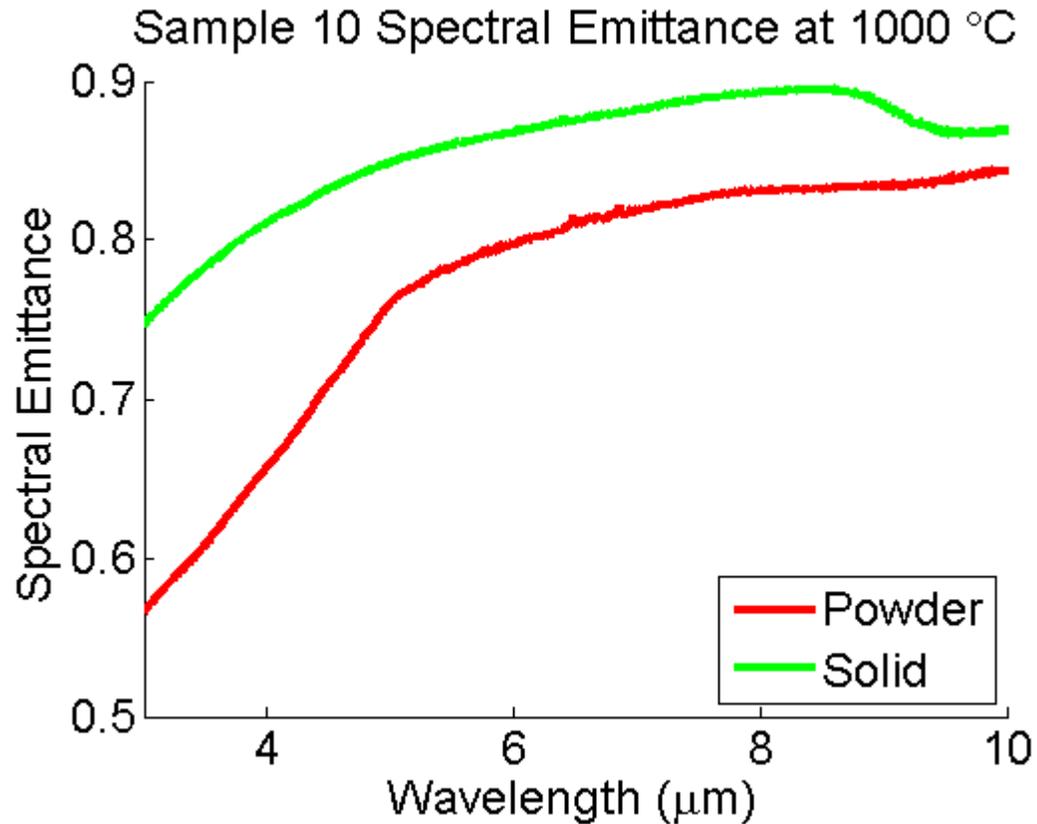
Powdery Ash Sample



Solid Slag Sample



Effect of Surface structure: Powder vs. Solid



- In general, coal slags (solids) have a higher emittance than coal ashes (powders)
- This is seen in both the spectral emittance and the total emittance

Conclusions

- Despite being from various locations in the furnace, the composition of all samples was very similar
 - Thus, no trend as a function of composition was distinguished
 - The change in emittance between sample location was contained within the changes between sample repetitions
- Spectral emittance did not change drastically (within 8%) in the temperature range examined (500-1000 °C)
- The spectral emittance did generally decrease with increasing temperature (as expected from the literature)
- Total emittance decreased (within 20%) in the temperature range examined (500-1000 °C)
 - The change in spectral emittance with temperature is amplified by weighting with Planck's distribution
- The surface structure (powder vs. solid) of the sample had a very significant effect on emittance
- The solid sample had significantly higher total emittance values (~20%) than the powdered sample



Acknowledgements

We acknowledge the support by the German Science Foundation (DFG) within the Sonderforschungsbereich/Transregio TR 129 “Oxyflame-Development of methods and models to describe solid fuel reactions within an oxyfuel-atmosphere“ for using the radiation test rig.

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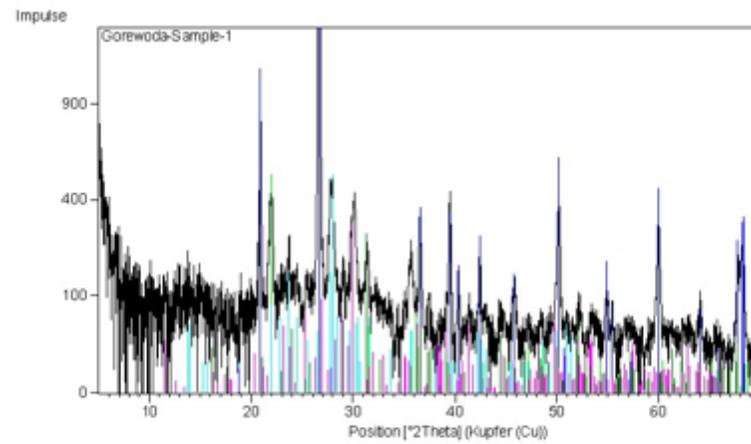
Thank you.



Supplemental Slides

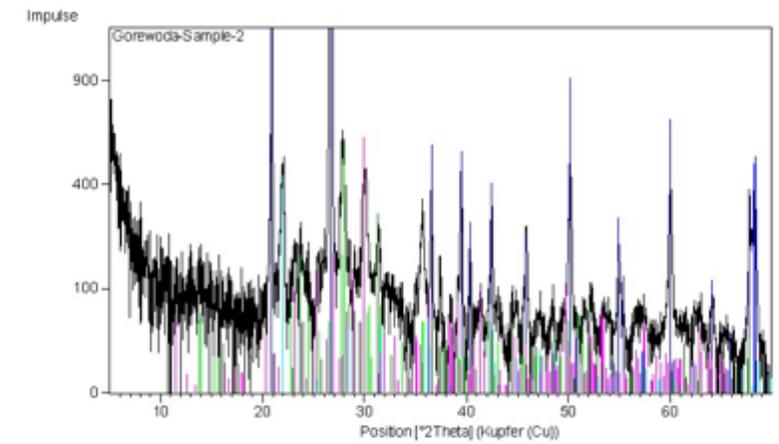
XRD

Diopside present in some samples may have formed in furnace.



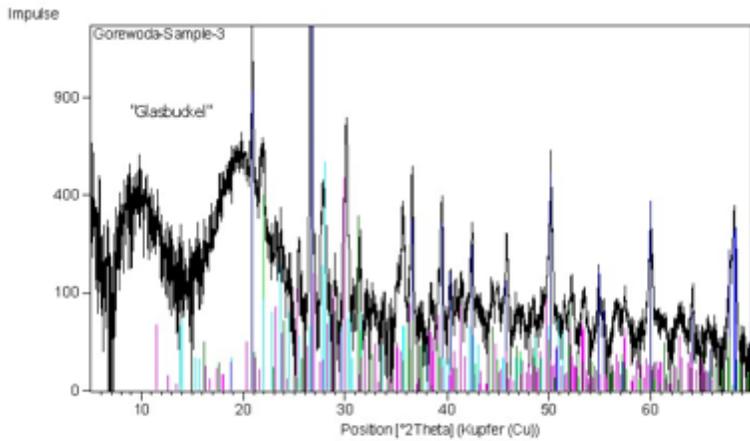
Reflexliste
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01-075-0923, Cristobalit (w. Si O2)
00-010-0393, Albit, NaAlSi3Al1O8
01-075-1399, Parawollastonit, CaSiO3
98-002-8358, Akemanzit-Celsidit, Al1.5Ca2Mg0.5O7Si1.5

Deutsches Bergbau-Museum Datei: Gorewoda-Sample-1 Datum: 20.07.2016 Dirk Kirchner



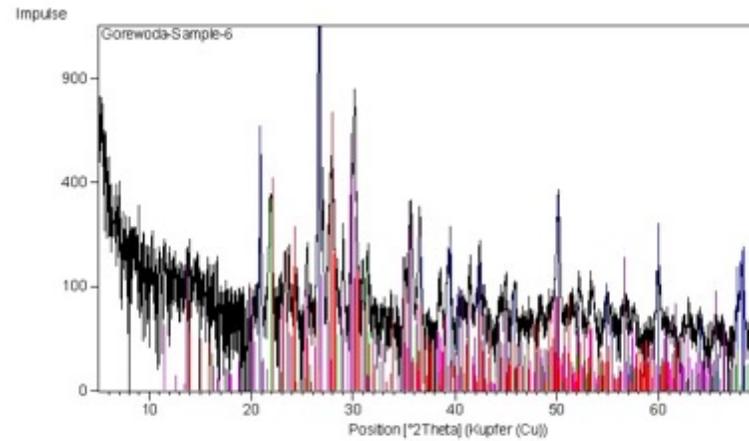
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01-075-1399, Parawollastonit, CaSiO3
98-002-8358, Akemanzit-Celsidit, Al1.5Ca2Mg0.5O7Si1.5

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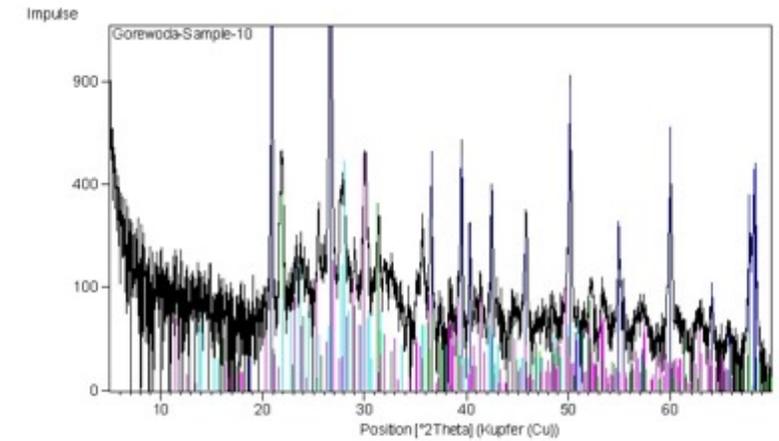
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Deutsches Bergbau-Museum Datei: Gorewoda-Sample-3 Datum: 20.07.2016 Dirk Kirchner



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01-075-1399, Parawollastonit, CaSiO3
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Deutsches Bergbau-Museum Datei: Gorewoda-Sample-6 Datum: 20.07.2016 Dirk Kirchner



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01-075-1399, Parawollastonit, CaSiO3
98-002-8358, Akemanzit-Celsidit, Al1.5Ca2Mg0.5O7Si1.5

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